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~~chemical name and~~

~~formula of some~~

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Chemical Formulas and

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SECTION 1 SHORT

ANSWER Answer the

following questions in the

space provided. 1. c In a

Stock system name such

as iron(III) sulfate, the

Roman numeral tells us

(a) how many atoms of

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Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

~~7 Chemical Formulas
and Chemical
Compounds~~

THE NOTORIOUS
B.I.C (BINARY IONIC
COMPOUND) •

Binary ionic compounds

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are compounds
composed of 2 different
elements. • As a general
rule, the total + charge
must ...

4

~~CHAPTER 7~~ ~~CHEMICAL~~ ~~COMPOUNDS &~~ ~~CHEMICAL~~ ~~FORMULAS~~

Ball-and-stick and space-
filling models show the
geometric arrangement

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of atoms in a molecule.

Isomers are compounds with the same molecular formula but different arrangements of atoms.

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Chemistry

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Chapter 7: Chemical
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Compounds Section 1:
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Formulas And
Chapter 7 Notes –
Chemistry; Chemical
Formulas and Chemical
Compounds. Molecules
and Molecular
Compounds. •

A molecule consists of two
or more atoms bounded
together. Molecules and
Chemical Formulas. •
Each molecule has a
chemical formula. •
The chemical formula

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~~Chapter 7 - Chemical~~

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Chapter 7: Chemical

Formulas and Chemical

Compounds Section 7-1:

Chemical Names and

Formulas 7-1-1 Explain

the significance of a

chemical formula.

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~~Compounds~~

Chapter Sixteen:

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Compounds 16.1
Chemical Bonds and
Electrons 16.2 Chemical
Formulas 16.3 Molecules
and Carbon Compounds
16.3 Molecules and
Carbon Compounds In
addition to the elements
from which it is made,
the shape of a molecule is
also important to its
function and properties.
We use structural
diagrams to show the

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shape and arrangement
of ...

Compounds

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Significance of a
Chemical Formula Recall
that a chemical formula
indicates the relative
number of atoms of each
kind in a chemical
compound.

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~~CHAPTER 7 Chemical Compounds~~ Formulas and Chemical Compounds

Ball-and-stick and space-filling models show the geometric arrangement of atoms in a molecule.

Isomers are compounds with the same molecular formula but different arrangements of atoms.

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Compounds. Chapter 7
is the final chapter of the
first semester.

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chapter and section numbers as they appear in the textbook. The Fill-in, Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples. The Common

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of chemistry, this book goes on exploring the concept of matter, energy, measurement, problem solving, atom, periodic table, and chemical bonding. These topics are followed by discussions on writing names and formulas of compounds; chemical formulas and the mole; chemical reactions; calculations based on

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equations; gases; and the properties of a liquid.

The remaining chapters examine the solutions;

acids; bases; salts;

oxidation-reduction

reactions;

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chemical kinetics and

equilibrium; and nuclear,

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guide will be of great

value to chemistry

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opportunities in
innovative,
nonconventional
materials and processes
that have the potential to
accelerate the
construction process,
improve the durability of
highway pavement and
bridges, and enhance the
serviceability and
longevity of new
construction under
adverse conditions.

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