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Chemical and Physical Changes

States of Matter : Solid Liquid Gas 10

Amazing Experiments with Water

Pure Substances and Mixtures!

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(Classification of Matter)Physical and
Chemical Changes

Understanding Atoms, elements, and
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and Change . Ch 1 – Introduction to
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Chemicals Ozone Layer, atmosphere,

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ozone formation, chlorofluorocarbons,
CFC's 1.2 Chemistry and Matter
Chemistry Central Science

**Chapter Summaries – Chemistry
Matter and Change**

Chapter 1 Section 2 MATTER and Its
Properties Matter – Anything that has

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MASS and VOLUME. ATOMS and MOLECULES the basic building blocks of matter. ATOM-- the smallest unit of an element that still maintains the properties that element. ELEMENT– a pure substance made up of only one type of atom.

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MATTER and CHANGE Chapter 1
Section 1

Chemistry: Matter and Change Chapter 6
Study Guide for Content Mastery . Name
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FOR CONTENT MASTERY Section 6.3
Periodic Trends In your textbook, read
about atomic radius and ionic radius.

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Circle the letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured ...

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Matter—Properties and

Changes Matter—Properties and Changes

Solutions Manual Chemistry: Matter and

Change • Chapter 3 35 Section 3.1

Properties of Matter pages 70–75 Problem-

Solving Lab 1. Explain why the flow of a

compressed gas must be controlled for

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practical and safe use.

**Matter—Properties and
Changes**
**Matter—Properties and
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Chemistry Greenhouse Effect, Movie
Ozone Studies, Movie Nylon, Movie Dr.

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Jekyll and Mr. Hyde, Movie Lab

Safety, Movie CD-ROM Chemistry:

Matter and Change Magic of

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multimedia for this chapter are available
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Chapter 1: Introduction to Chemistry

Page 19/35

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Guide and Answers 7 Study Guide -

Chapter 6 – The Periodic Table and

Periodic Law Section 6.1 Development of

the Modern Periodic Table 1. octaves 2.

eight 3. nine 4. accepted 5. Dmitri

Mendeleev 6. atomic mass 7. elements 8.

atomic number 9. Henry Moseley 10.

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protons 11. periodic law 12. properties 13.
14 ...

Ch 6 Study Guide answers

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5 25 Electrons in Atoms Reviewing

Vocabulary Match the definition in

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Column A with the term in Column B.
Column A Column B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

Chapter Assessment

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used in chemistry can be q... the study of
energy changes (that is, energy produced
or absor... A system that can exchange
neither energy nor matter with its...

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Chemistry: Chapter 2: Matter and Change.

extensive property. intensive property.

extensive properties (examples) intensive

property (examples) a property that

depends on the amount of matter in a

sample. a property that depends on the

type of matter in a sample not.... mass,

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volume, length, weight.

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Math Skills Transparency Masters

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11 31 number of moles of known

substance C H C H C H O O O CO CO

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CO H O H O H O number of moles of
unknown substance O CO H O C H CO H
O C H O H O C H O CO use mole ratio 5
mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol
C 3 H 8 4 mol H 2 O 1 mol C 3 H 8 1 mol
C 3 H 8 5 mol O 2

MATH SKILLS TRANSPARENCY

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A “chemical change” is a change that produces matter with a different composition than the original matter.

Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

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Properties of Compounds Quite different
properties than their component elements.

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