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overwhelm

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students and
force them to
fall back on
memorization.
But once they
understand how
to use
mechanisms, they
can solve just
about any

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problem. With an organization by mechanism, students will understand more, and memorize less.

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chemistry was
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traditionally

defined as the
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While organic

chemistry opens

the door to the

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not limited to
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chemistry is
also the basis
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and computers to
name a few.

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Organic chemists use a curved arrow (\curvearrowright) to show the movement of an electron pair. A sigma (σ) bond is formed between atoms by the overlap of two atomic orbitals along the line that connects the

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atoms. Carbon

uses

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sp³-hybridized
orbitals to form

four such bonds.

These bonds are
directed from

the carbon

nucleus toward

the corners of a

tetrahedron.

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new reactions,

and

extraordinary
syntheses have

been manifold.

New techniques

and new

instruments for

analysis and

determination of

structures,

improved methods

for theoretical

calculations, as

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well as new

junctions with

physical,

inorganic, and

biochemistry,

have made

organic

chemistry an

enormously vital

discipline.

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OF ORGANIC

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The goal is not to understand every single thing about this organic chemistry mechanism in great detail right now. Our goal right now is just to appreciate how nucleophiles and

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electrophiles

are used in

organic

chemistry

mechanisms and

start to get a

feeling for how

these curved

arrows show the

movement or flow

of electrons.

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